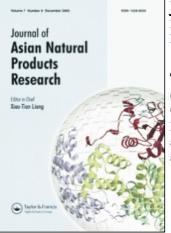
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## Three new triterpenes from the roots of Rhus javanica L. var. roxburghiana

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Three new triterpenes, lantabetulal (1), lantanolal (2), and lantanolol (3) along with lantabetulic acid (4) and lantanolic acid (5) were isolated from the roots of *Rhus javanica* L. var. *roxburghiana*. The structural assignments of all these compounds were carried out by their spectroscopic analysis.

Keywords: Rhus javanica L. var. roxburghiana; Anacardiaceae; lantabetulal; lantanolal; lantanolol

#### 1. Introduction

Rhus javanica L. var. roxburghiana, a smallsized deciduous tree, is widely distributed in thickets and secondary forests at low altitudes throughout East Asia including India, China, Taiwan, Korea, and Japan [1]. Its roots have been used in folk medicines as refrigerant and anti-parasitic agents, and for the treatments of malaria and respiratory organ-related diseases [2]. Previous chemical examinations on R. *javanica* afforded flavonoids [3,4], triterpenoids [5-7], phenolics [3,8], one tannin [9], and one aromatic alkane [5] from its stem, bark, leaves, gall, and adventitious root cultures. However, there has been no report on the constituents of the roots of R. javanica L. var. roxburghiana. In the course of investigations into the chemical constituents of R. javanica L. var. roxburghiana root, two lupane triterpenes 1 and 4 accompanied with three oleanane triterpenes 2, 3, and 5 were isolated and identified from the roots of this plant. We now report the structural assignments of these compounds.

## 2. Results and discussion

Chemical investigations on the roots of *R. javanica* L. var. *roxburghiana* afforded three new triterpenes 1-3 and two known analogues, lantabetulic acid (4) [10] and lantanolic acid (5) [11]. The structures of 1-3 (Figure 1) were established mainly by <sup>1</sup>H, <sup>13</sup>C NMR and DEPT spectra and 2D-NMR experiments including HMQC, HMBC, and NOESY, and the results of NMR assignments were further confirmed by the data of mass spectrometry and FT-IR spectroscopy.

Compound **1** was assigned a molecular formula of  $C_{30}H_{46}O_3$ , as deduced from the HR-EI-MS, <sup>1</sup>H, and <sup>13</sup>C NMR spectral data. Its IR spectrum indicated the presence of a hydroxyl group (3395 cm<sup>-1</sup>) and a carbonyl

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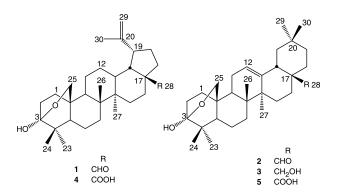


Figure 1. Structures of compounds 1-5.

group  $(1724 \text{ cm}^{-1})$ . The <sup>1</sup>H NMR spectrum (Table 1) of 1 showed signals due to four tertiary methyl groups at  $\delta_{\rm H}$  0.80, 0.94, 0.95, and 1.00 two olefinic protons at  $\delta_{\rm H}$  4.62 (1H, br s) and 4.74 (1H, br s), two oxygen-bearing methylene protons at  $\delta_{\rm H}$  3.70 (1H, dd, J = 1.6, 8.4 Hz) and 4.21 (1H, dd, J = 2.4,8.4 Hz), and an aldehyde functionality at  $\delta_{\rm H}$ 9.63 (1H, br s). The carbon signals observed in the <sup>13</sup>C NMR and DEPT spectra of 1 suggested the presence of one aldehyde carbonyl group at  $\delta_{\rm C}$  206.4, one 1,1disubstituted double bond at  $\delta_{\rm C}$  110.2 and 149.6, one oxygen-bearing methylene carbon at  $\delta_{\rm C}$  68.0, and one hemiacetal carbon at  $\delta_{\rm C}$ 98.0. On account of the molecular formula  $C_{30}H_{46}O_3$ , the degree of unsaturation of 1 was eight including one aldehyde carbonyl and one olefinic functionality. Thus, the number of rings of 1 should be six. The COSY spectrum of 1 showed that the two oxygenbearing methylene protons at  $\delta_{\rm H}$  3.70 and 4.21 had W-shaped <sup>1</sup>H-<sup>1</sup>H correlations with those of  $H_{\alpha}$ -1 and H-5, respectively. The HMQC and HMBC spectra of 1 showed that two oxymethylene protons at  $\delta_{\rm H}$  3.70 and 4.21 had long-range correlations with C-3 ( $\delta_{\rm C}$ 98.0) and C-5 ( $\delta_{\rm C}$  49.8). All above spectroscopic data suggested that 1 was closely comparable to a lupane-type triterpene aldehyde skeleton except for that H<sub>3</sub>-25 singlet was converted to an oxymethylene functionality that bridged between C-3 and C-25, the same as in 4. The NOESY spectrum of

1 exhibited mutual correlations between  $H_{a}$ -25,  $H_{b}$ -25, and  $H_{3}$ -26 (Figure 2), confirming  $H_{2}$ -25 to be axial oriented. Further analysis of all the 2D-NMR spectral data allowed the complete assignments of <sup>1</sup>H and <sup>13</sup>C NMR spectra of 1, as illustrated in Table 1. Accordingly, compound 1 was concluded to be 3 $\beta$ ,25-epoxy-3-hydroxylupan-28-al, and named lantabetulal.

Compound 2 was obtained as colorless needles. Its molecular formula C<sub>30</sub>H<sub>46</sub>O<sub>3</sub> was deduced from the  $[M]^+$  ion at m/z 454.3440 in the HR-EI-MS. Analysis of the IR spectrum of 2 suggested that it contained a hydroxyl group  $(3540 \,\mathrm{cm}^{-1})$  and a carbonyl group  $(1724 \text{ cm}^{-1})$ . The <sup>1</sup>H NMR signals of 2 included six singlet methyl groups at  $\delta_{\rm H}$  0.64, 0.88, 0.89, 0.94, 1.00, and 1.12, one trisubstituted olefinic proton at  $\delta_{\rm H}$  5.35 (dd, J = 3.6, 3.6 Hz, H-12), and an aldehyde functionality at  $\delta_{\rm H}$  9.35 (1H, s) at low-field region (Table 1). The <sup>13</sup>C NMR and DEPT spectral data showed an oxygenated and dioxygenated carbons ( $\delta_C$  67.7 and 98.0), a trisubstituted double bond ( $\delta_{\rm C}$  123.2 and 142.7), and a carbonyl functional group ( $\delta_{\rm C}$ 207.1). Based on the above data together with a base peak of EI-MS at m/z 203, compound 2 was deduced to be an oleanene-type triterpene aldehyde skeleton with one aldehyde group at C-17, and one double bond at C-12. Its COSY, NOESY, and HMBC correlations also indicated a β-oriented oxygen-bearing methylene spin system located at C-25, the same as

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|-------------|------|----|-----|
|             |      |    |     |

|     |                  | 1   |                  | 2   |                  | 3   |  |
|-----|------------------|---|------------------|---|------------------|---|--|
| No. | $\delta_{\rm C}$ | $\delta_{\rm H}$ mult. $(J/{\rm Hz})^{\rm a}$ | $\delta_{\rm C}$ | $\delta_{\rm H}$ mult. $(J/{\rm Hz})^{\rm a}$ | $\delta_{\rm C}$ | $\delta_{\rm H}$ mult. $(J/{\rm Hz})^{\rm a}$ |  |
| 1   | 29.9             | 1.65  | 29.5             | 1.68  | 29.6             | 1.14  |  |
|     |                  | 2.16  |                  | 2.10  |                  | 1.69  |  |
| 2   | 35.2             | 1.08  | 34.9             | 1.16  | 35.0             | 1.12  |  |
|     |                  | 2.14  |                  | 2.09  |                  | 1.18  |  |
| 3   | 98.0             |   | 98.0             |   | 98.0             |   |  |
| 4   | 35.6             |   | 34.9             |   | 35.0             |   |  |
| 5   | 49.8             | 1.18  | 50.4             | 1.18  | 50.4             | 1.18  |  |
| 6   | 19.7             | 1.42  | 19.5             | 1.39  | 19.7             | 1.18  |  |
|     |                  | 1.48  |                  | 1.48  |                  | 1.50  |  |
| 7   | 32.1             | 1.30  | 31.2             | 1.35  | 31.0             | 1.29  |  |
|     |                  | 1.42  |                  | 1.42  |                  | 1.39  |  |
| 8   | 39.8             |   | 38.5             |   | 38.8             |   |  |
| 9   | 45.1             | 1.40  | 41.8             | 1.59  | 41.9             | 1.64  |  |
| 10  | 40.4             |   | 40.1             |   | 40.2             |   |  |
| 11  | 22.0             | 1.28  | 23.7             | 1.90  | 23.8             | 1.90  |  |
|     |                  | 1.54  |                  | 1.98  |                  | 1.94  |  |
| 12  | 25.8             | 1.72  | 123.2            | 5.35 dd (3.6, 3.6)                            | 122.3            | 5.21 t (2.8)                                  |  |
|     |                  | 1.78  |                  |   |                  |   |  |
| 13  | 38.9             | 2.04  | 142.7            |   | 144.1            |   |  |
| 14  | 42.3             |   | 41.8             |   | 41.8             |   |  |
| 15  | 29.0             | 1.38  | 26.7             | 1.56  | 25.5             | 1.06  |  |
|     |                  | 2.08  |                  | 1.60  |                  | 1.30  |  |
| 16  | 28.9             | 1.20  | 22.1             | 1.54  | 22.1             | 1.18  |  |
|     |                  | 1.39  |                  | 1.94  |                  | 1.85  |  |
| 17  | 59.3             |   | 49.2             |   | 37.1             |   |  |
| 18  | 47.8             | 1.74  | 40.9             | 2.61 dd (4.4, 14.0)                           | 42.8             | 1.98  |  |
| 19  | 47.3             | 2.85 dt (5.6, 10.8)                           | 45.3             | 1.18  | 46.2             | 1.06  |  |
|     |                  |   |                  | 1.67  |                  | 1.72  |  |
| 20  | 149.6            |   | 30.6             |   | 30.9             |   |  |
| 21  | 29.6             | 1.36  | 27.6             | 1.20  | 34.1             | 1.16  |  |
|     |                  | 1.40  |                  | 1.37  |                  | 1.28  |  |
| 22  | 33.1             | 1.62  | 33.1             | 1.02  | 30.9             | 1.24  |  |
|     |                  | 1.74  |                  | 1.24  |                  | 1.52  |  |
| 23  | 26.9             | 1.00 s  | 27.2             | 1.00 s  | 27.2             | 1.03 s  |  |
| 24  | 18.3             | 0.95 s  | 18.2             | 0.94 s  | 18.2             | 0.97 s  |  |
| 25  | 68.0             | 3.70 dd (1.6, 8.4)                            | 67.7             | 3.72 dd (1.6, 8.8)                            | 67.9             | 3.88 dd (1.6, 8.8)                            |  |
|     |                  | 4.21 dd (2.4, 8.4)                            |                  | 4.21 dd (2.8, 8.8)                            |                  | 4.26 dd (2.8, 8.8)                            |  |
| 26  | 16.2             | 0.80 s  | 17.5             | 0.64 s  | 17.2             | 0.88 s  |  |
| 27  | 13.9             | 0.94 s  | 24.9             | 1.12 s  | 25.3             | 1.15 s  |  |
| 28  | 206.4            | 9.63 br s                                     | 207.1            | 9.35 s  | 69.9             | 3.19 d (10.8)                                 |  |
| 20  | 110.0            | 4 (0)   | 22.0             | 0.00  | 22.1             | 3.51 d (10.8)                                 |  |
| 29  | 110.2            | 4.62 br s                                     | 33.0             | 0.89 s  | 33.1             | 0.88 s  |  |
| 20  | 10.1             | 4.74 br s                                     | 22.2             | 0.00  | 22.5             | 0.07  |  |
| 30  | 19.1             | 1.68 s  | 23.3             | 0.88 s  | 23.5             | 0.86 s  |  |

Table 1. <sup>1</sup>H and <sup>13</sup>C NMR spectral data for compounds **1**, **2**, and **3** (CDCl<sub>3</sub>, 500 MHz).

<sup>a</sup> Signals without multiplicity were picked up from COSY or HMQC spectra.

in 1. When comparing the spectral data of 2 with those of 5, the difference is only an aldehyde functionality in 2 instead of a carboxylic acid group at C-17 in 5. Thus, compound 2 was established as  $3\beta$ ,25-epoxy-

3-hydroxyolean-12-en-28-al, and we called it lantanolal.

Compound 3 possessed spectroscopic data closely comparable to those of 2 except that its aldehyde functionality in 2 was

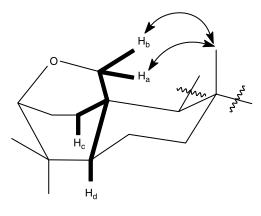


Figure 2. Key NOESY correlations (  $\leftrightarrow$  ) of **1**.

replaced with a hydroxymethylene group in 3. Its <sup>1</sup>H NMR (Table 1) spectrum exhibited signals for additional two oxygen-bearing methylene protons at  $\delta_{\rm H}$  3.19 and 3.51 (d, J = 10.8 Hz), not the aldehyde protons ( $\delta_{\rm H}$ 9.35, s) observed in 2. Hence, 3 contained a hydroxymethylene moiety instead of an aldehyde, which was also supported by the evidence that both H<sub>a</sub>-28 and H<sub>b</sub>-28 correlated with one quaternary carbon (C-17,  $\delta_{\rm C}$ 37.1) and one secondary carbon (C-22,  $\delta_{\rm C}$ 30.9) in the HMBC spectrum of 3. The peaks of EI-MS in 3 at m/z 234 (16%) and 203 (100%) also indicated a 12-oleanene with a hydroxymethylene at C-17 [12]. Therefore, 3 was assigned as 3B,25-epoxy-3-hydroxyolean-12-en-28-ol, and named lantanolol.

#### 3. Experimental

## 3.1 General experimental procedures

Melting points were collected using a Yanaco MP-53 apparatus and are uncorrected. Optical rotations were measured with JASCO DIP-180 digital polarimeter at room temperature. IR spectra were recorded on a Perkin-Elmer 983G spectrophotometer. <sup>1</sup>H and <sup>13</sup>C NMR spectra were recorded on a Bruker AM-300 and DMX-500 instruments using tetramethyl-silane (TMS) as an internal standard. Chemical shifts are given in  $\delta$  values (ppm) and coupling constants (*J*) are given in hertz (Hz). EI-MS was obtained on JMS-HX 110 mass spectrometer. For thin-layer chromatography

analysis, silica gel 60  $F_{254}$  plates (Merck, Germany) were used. HPLC was performed on GBC LC-1440 and LDC Analytical-III liquid chromatographs with Lichrosorb Si-60 column (7  $\mu$ m, 250 × 10 mm, Merck, Germany).

#### 3.2 Plant material

The roots of *R. javanica* L. var. *roxburghiana* were collected at Kaohsiung, Taiwan, in June 2000. The plant material was identified by Mr Muh-Tsuen Gun, a former technician in the Department of Botany, National Taiwan University. A voucher specimen (YHK071020) has been deposited at the Herbarium of the Department of Botany, National Taiwan University, Taipei, Taiwan.

#### 3.3 Extraction and isolation

The air-dried roots of R. javanica L. var. roxburghiana (18.0 kg) were extracted with 1201 MeOH thrice (7 days each time) at room temperature. The combined extracts were evaporated under vacuum to give a residue (1050 g) that was suspended in 81 water and then partitioned with EtOAc and n-BuOH successively. The ethyl acetate layer was then chromatographed by Si-column and HPLC repeatedly. The eluent systems are combinations of n-hexane, EtOAc, and MeOH in a stepwise elution mode. Fractions eluted by EtOAc/hexane (3:7) were further purified by repetitive HPLC with EtOAc/hexane (3:7) as eluent to afford 1 (5 mg), 2 (20 mg), 4 (6 mg), and 5 (9 mg). Fractions eluted by EtOAc/hexane (1:1) were further purified by HPLC using the same column with EtOAc/hexane (1:1) as eluent to afford 3 (6 mg).

#### 3.3.1 Lantabetulal (1)

Colorless needles; mp 153–155°C;  $[\alpha]_D^{24}$  + 49.4 (*c* 0.14, CHCl<sub>3</sub>); IR  $\nu_{max}$  (neat) 3395, 2926, 1724, 1457, 1378 cm<sup>-1</sup>; <sup>1</sup>H and <sup>13</sup>C NMR spectral data, see Table 1; EI-MS *m/z* (rel. int. %) 454 (M<sup>+</sup>, 100), 425 (26), 407 (36),

219 (17), 189 (25); HR-EI-MS m/z 454.3440 [M]<sup>+</sup> (calcd for C<sub>30</sub>H<sub>46</sub>O<sub>3</sub>, 454.3449).

## 3.3.2 Lantanolal (2)

Colorless needles; mp 145–147°C;  $[\alpha]_D^{24}$  + 95.0 (*c* 2.79, CHCl<sub>3</sub>); IR  $\nu_{max}$  (neat) 3450, 1724, 1466, 1383 cm<sup>-1</sup>; <sup>1</sup>H and <sup>13</sup>C NMR spectral data, see Table 1; EI-MS *m/z* (rel. int. %) 454 (M<sup>+</sup>, 98), 425 (48), 407 (42), 232 (27), 203 (100), 189 (44), 145 (72); HR-EI-MS *m/z* 454.3440 [M]<sup>+</sup> (calcd for C<sub>30</sub>H<sub>46</sub>O<sub>3</sub>, 454.3449).

### 3.3.3 Lantanolol (3)

Colorless plates; mp 128–129°C;  $[\alpha]_D^{24}$  + 128.0 (*c* 0.52, CHCl<sub>3</sub>); IR  $\nu_{max}$  (neat) 3404, 1648, 1465, 1365 cm<sup>-1</sup>; <sup>1</sup>H and <sup>13</sup>C NMR spectral data, see Table 1; EI-MS *m/z* (rel. int. %) 456 (M<sup>+</sup>, 22), 425 (21), 407 (8), 255 (42), 234 (10), 203 (100), 189 (22), 145 (42); HR-EI-MS *m/z* 456.3576 [M]<sup>+</sup> (calcd for C<sub>30</sub>H<sub>48</sub>O<sub>3</sub>, 456.3605).

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